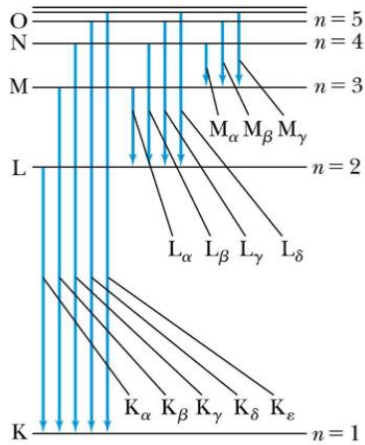


1. Moseley
2. It tells you the ratio of the velocity of n=1 electron to the speed of light
3. He assumed light was absorbed and emitted in quanta.
4. a decrease in the necessary stopping voltage.
5. Electrons transitioning down from an outer shell replace electrons ejected from an inner shell
6. $P = 180 \text{ W}$, $P_\gamma = 4 \text{ MeV}$

$$\frac{180 \text{ J}}{1 \text{ s}} * \frac{1 \text{ MeV}}{1.6 \times 10^{-13} \text{ J}} * \frac{1 \text{ Photon}}{4 \text{ MeV}} = \underline{2.81 \times 10^{14} \text{ photon/s}}$$

7. From the O-shell to a vacancy in the M-shell.



8. $T = 7562^\circ\text{C} = 7835\text{K}$, $\text{const} = 2.9 \times 10^{-3} \text{ (m}^*\text{K)}$

$$\lambda = \frac{\text{const}}{T} = \frac{2.9 \times 10^{-3} \text{ m}^*\text{K}}{7835 \text{ K}} = 3.7 \times 10^{-7} \text{ m} = \underline{370 \text{ nm}}$$

9. Be^{+++}

10. excitation of a Hg atom to the first excited state.

11. $B = 0.032$, $\lambda = 607 \text{ nm}$, $\lambda = \frac{c}{f}$

Note: The signs are flipped because we converted from frequency to wavelength.

$$\lambda' = \lambda \sqrt{\frac{1-\beta}{1+\beta}} = 607 \sqrt{\frac{1-0.032}{1+0.032}} \approx \underline{588 \text{ nm}}$$

12. $a_0 = 0.5 \times 10^{-10} \text{ (m)}$, $r_p = 0.875 \times 10^{-15} \text{ (m)}$

$$r_2 = n^2 a_0 = 4 * 0.5 \times 10^{-10} \text{ (m)} = 2 \times 10^{-10} \text{ (m)}$$

$$\frac{r_2}{r_p} = \frac{2E - 10 (m)}{0.875E - 15 (m)} = 2.3E5$$

$$0.01(m) * 2.3E5 = \underline{2300 (m)}$$

13. All wavelength's scatter the same

14. The nucleus and electron(s) revolve around their mutual center of mass which virtually decreases the apparent mass of the electron.

15. $v_0 = 0.5c$

$$P_0 = \gamma_0 m v_0 = \frac{1}{\sqrt{1-0.5^2}} * m * (0.5c) = m * 0.577c$$

$$P_1 = 1.5 * P_0 = 0.866c$$

$$v_1 = \frac{P_1 * c}{\sqrt{1+P_1^2}} = \underline{0.655c}$$

16. The emission of a photon from an electron being accelerated by a nucleus.

17. $r_n = n^2 a_0$, $a_0 = 0.5E-10(m)$

$$r_6 = 6^2 * 0.5E-10(m) = 18E-10(m)$$

$$D = 2 * r = 2 * 18E-10(m) = \underline{36E-10(m)}$$

18. $T = \frac{const}{\lambda}$, $const = 2.9E-3 (m * K)$

$$\text{for } \lambda = 9348(nm), T \approx 310K \approx 38^\circ C$$

19. $m_p = 938 (MeV/c^2)$, $m_\tau = 1777 (MeV/c^2)$, $k = \frac{1}{4\pi\epsilon}$, $k * e^2 = 1.44E-15(MeV * m)$, $\hbar c = 197 (MeV * fm)$

$$\mu = \frac{m_p m_\tau}{m_p + m_\tau} = 613.9(MeV/c^2)$$

$$a_0 = \frac{\hbar^2}{k * \mu * e^2} = \underline{1711.3 (fm)}$$

Note: for this problem, keeping track of units is very important. Using the already precompiled constants and unit conversions located on the tables at the front of the textbook is helpful.

20. $\lambda = 500 (nm)$, $R_{earth} = 1400(W/m^2)$, $A = 1(m^2)$, $\epsilon = 20\%$, $\Phi_{bc} = 1.49(eV)$

$$P_{delivered} = R_{earth} * A = 1400(W)$$

$$P_{electrical} = \epsilon * P_{delivered} = 0.2 * 1400(W) = 280(w) \text{ \& } P_{electrical} = V * I$$

$$V = hf - \Phi_{bc} = \frac{1240(eV * nm)}{500(nm)} - 1.49(eV) = \underline{0.99(V)}$$

$$I = \frac{P_{electrical}}{V} = \frac{280(W)}{0.99(V)} = \underline{282.8(A)}$$